Molarity	no of moles of solute in 1 litre of solution
Molality	no of moles of solute in 1 kg solvent
Azeotropes	show positive deviation
SPM	semipermeable membrane
Normality	no of gram equivalent of solute in 1 Litre solution
Ideal solution	obey Roults law
Colligative property	depend on no of solute
Vant Hoff factor	normal molar mass \ observed molar mass
Nonideal solution	not obey Roult Law
ppm	parts per million

$mole\,fraction$	$ratio\,of\,moles\,of\,solute\,to\,total\,moles$
$solubility$	$amount\,of\,solute\,in\,100\,gm\,solvent$
$Liquid * liquid$	$aq\,alcohol\,solution$
$Gas * Gas$	air
$Henry\,law$	$solubility\,of\,gas\,in\,solution$
$mass\%$	$mass\,of\,solute * 100 \mid mass\,of\,solution$
$solution$	$homogeneous$
$osmosis$	$movement\,of\,solvent\,through\,SPM$

Molarity

amount of solute in 100 gm solvent

Molality

no of moles of solute in 1 litre of solution

solubility of gas in solution

Gas * Gas

homogeneous

obey Roults law

aqalcohol solution

Liquid * liquid

dependonnog solute

Vant Hoff factor

udd

ppm

not obey Roult Law

ratio of moles of solute to total moles

Azeotropes

Normality

semipermeablemembrane

movementofsolventthroughSPM

Nonidealsolution

normalmolarmass\observedmolarmass

massofsolute*100|massofsolution

solubility

parts permillion

solution

mole fraction

SPM

Idealsolution

Colligativeproperty

noofgramequivalentofsolutein1Litresolution

osmosis

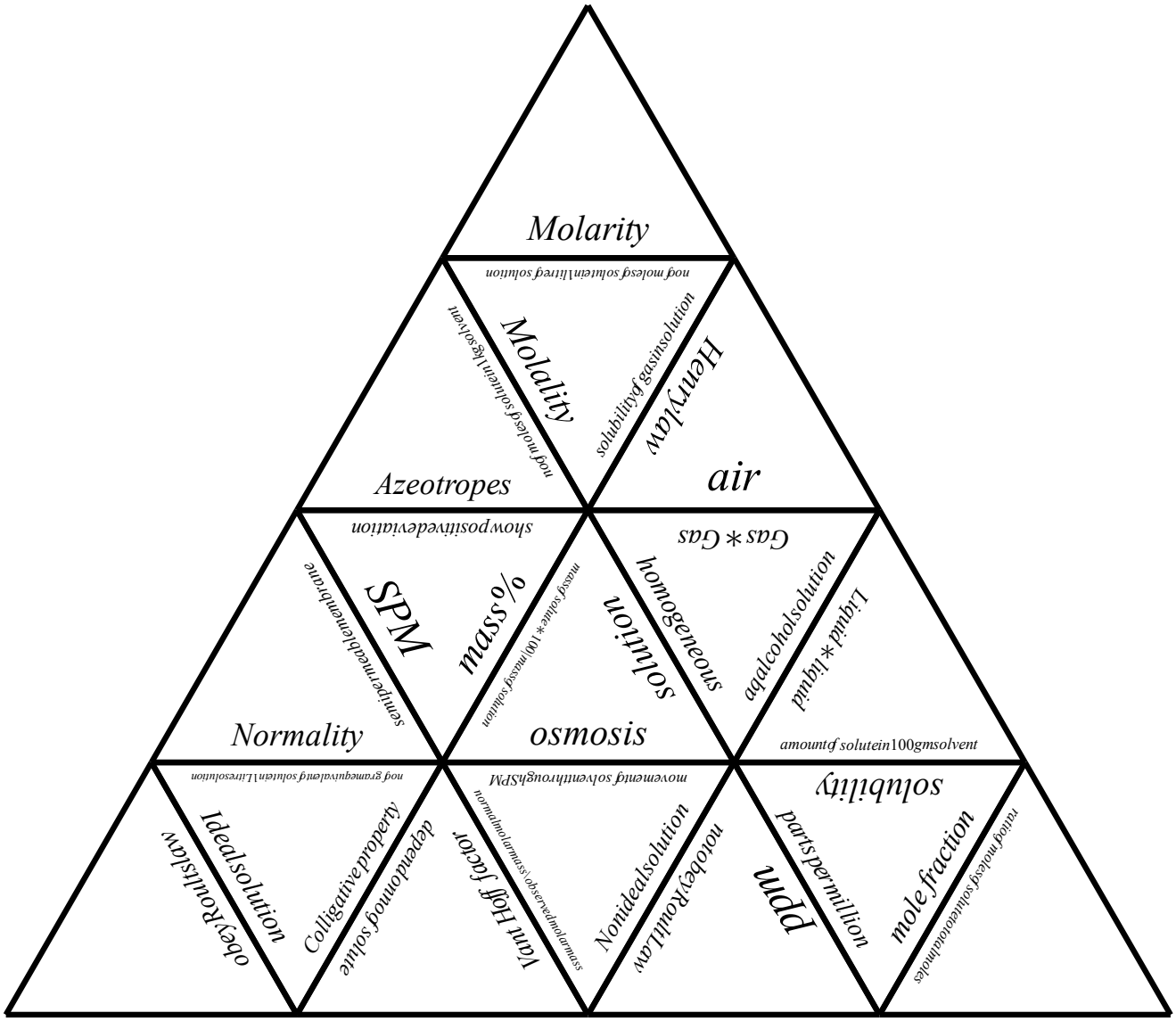
air

Henrylaw

mass%

showpositivedeviation

noofmolesofsolutein1kgsolvent


Molarity
mol moles of solute/L litre of solution
Molality
mol moles of solute/kg solvent
Henry's law
solubility of gas in solution
Azeotropes
air
show positive deviation
SPM
sent ppe metal element ratio
mass %
mass of solute * 100/mass of solution
solution
homogeneous
Gas * Gas
aqa lcohol solution
Liquid * liquid
Normality
osmosis
amount of solute in 100 gm solvent
mol g gram equivalent g solute/L litre solution
movement of solvent through SPM
Ideal solution
obey Roult's law
Colligative property
depend on mol g solute
Van Hoff factor
normal molar mass/observed molar mass
Nonideal solution
not obey Roult's law
ppm
parts per million
solubility
mole fraction
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